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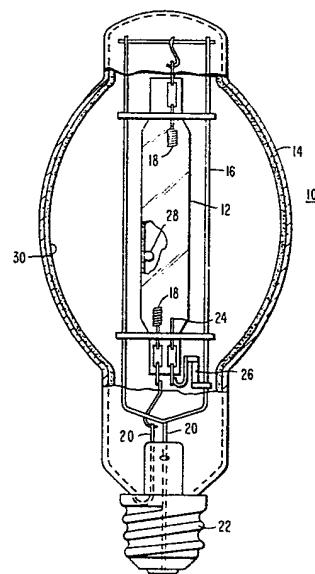
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㉓ Light source for illuminating objects with enhanced perceived coloration.

㉔ Light sources modified to provide enhanced perceived coloration of illuminated objects. Light sources as normally designed, without the present modification, have an envelope-enclosed source of radiations having appreciable emissions in the wavelength range of from 565 nm to 595 nm plus substantial emissions of visible radiations of wavelengths shorter than 565 nm. A phosphor is carried on the surrounding envelope and is operable to absorb radiations in the wavelength range of from 565 nm to 595 nm and to convert absorbed radiations to longer wavelength emission in the range of from 596 nm to 640 nm.



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LIGHT SOURCES FOR ILLUMINATING OBJECTS

This invention relates to light sources for illuminating objects and, more particularly, to a light source which is modified to provide enhanced perceived coloration of illuminated objects.

5 In Journal of the Optical Society of America, Volume 61, No. 9, pp. 1155-1163 (September 1971), article entitled "Luminosity and Color-Rendering Capability of White Light" by W. A. Thornton, it is disclosed that maximum luminous efficacy and color rendering are achieved by 10 blending three spectral lines at about 450 nm, 540 nm and 610 nm. Also the wavelength regions near 500 nm and 580 nm are shown to be disadvantageous with respect to obtaining good color rendering.

15 In Journal of the Optical Society of America, Volume 62, No. 3, pp. 457-459 (March 1972), article entitled "Three-Color Visual Response" by W. A. Thornton, it is disclosed that various experiments indicate that three-color response of human vision consists of well-separated channels peaking near 450 nm, 540 nm and 610 nm.

20 In Lighting Design and Application, pp. 51 and 52, December 1972, article entitled "The Quality of White Light" by W. A. Thornton, it is disclosed that by mixing three wavelengths of about 450 nm, 540 nm and 610 nm, colors of illuminated objects can be rendered in a preferential fashion.

25 U.S. Patent No. 2,748,303 (Thorington) discloses a phosphor coating carried on the inner surface of the outer protective envelope for a high-pressure mercury-

vapor discharge lamp, which phosphor corrects the color output of the composite lamp.

U.S. Patent No. 3,602,758 (Thornton and Sausville) discloses a dual phosphor coating which reduces the 5 proportions required for more costly phosphors.

According to the present invention a light source for illuminating objects with an enhanced perceived coloration comprises an electrically energizable source of radiations surrounded by a light-transmitting envelope, 10 said source of radiations when energized having appreciable emissions in the wavelength range of from 565 nm to 595 nm plus substantial emissions of visible wavelengths shorter than 565 nm; characterized in that a layer comprising inorganic phosphor is carried on the inner surface 15 of the light-transmitting envelope, said phosphor being absorptive for radiations in the wavelength range of from 565 nm to 595 nm and transmissive for visible radiations shorter than 565 nm and longer than 595 nm, and said phosphor layer being able to convert a substantial proportion 20 of absorbed radiations in the range of from 565 nm to 595 nm to longer wavelength radiations in the wavelength range of from 596 nm to 640 nm. In this manner, yellow-wavelength appearing radiations are converted to orange to 25 red radiations and the perceived coloration of illuminated objects is enhanced.

In order that the invention can be more clearly understood, convenient embodiments thereof will now be described, by way of example, with reference to the accompanying drawings in which:

30 Fig. 1 is an elevational view, partly in section, showing a high-pressure mercury-vapor lamp the outer envelope of which is coated with a particular phosphor material;

35 Fig. 2 is the spectral power distribution for a bare or uncoated HID mercury lamp showing the lamp emission for the wavelength range of from 540 nm to 590 nm;

Fig. 3 is the spectral power distribution for the wavelength range of from 570 nm to 670 nm for a par-

ticular phosphor which may be used in accordance with the present invention;

5 Fig. 4 is a graph showing the excitation spectrum for the wavelength range of from 569 nm to 588 nm for the same phosphor whose emission spectrum is shown in Fig. 3;

10 Fig. 5 is a graph of color rendering index versus wavelength showing the color rendering index which is obtained from white light composed of three selected emissions of varying wavelengths;

Fig. 6 is a graph of relative intensity versus wavelength wherein the standard photopic curve is shown as a dashed line and a modified eye-response curve is shown as a solid line;

15 Fig. 7 is a fragmentary enlarged sectional view of a portion of the outer envelope of an HID mercury lamp, such as is shown in Fig. 1, but illustrating an alternative phosphor coating;

20 Fig. 8 is an enlarged fragmentary sectional view similar to Fig. 7 showing yet another alternative phosphor coating;

Fig. 9 is an elevational view, shown partly in section, of a high-intensity-discharge sodium lamp modified in accordance with the present invention;

25 Fig. 10 is an elevational view, shown partly in section, of an HID mercury-metal halide lamp modified in accordance with the present invention;

30 Fig. 11 is an elevational view, shown partly in section, of a fluorescent lamp modified in accordance with the present invention; and

Fig. 12 is an elevational view, shown partly in section, of a standard incandescent-type lamp modified in accordance with the present invention.

35 Referring to Fig. 1, high-intensity-discharge (HID) lamp 10 comprises an arc tube 12 supported within a light-transmitting protective outer envelope 14 by a conventional supporting frame 16. Electrodes 18 are operatively supported within the arc tube proximate the ends

thereof and are electrically connected through lead-in conductors 20 to a conventional screw-type base 22 for connection to a source of power. One of the electrodes has a starting electrode 24 positioned proximate thereto 5 which is electrically connected through a resistor 26 to the source of power in order to facilitate starting. The arc tube contains a conventional charge of mercury 28 and a small charge of inert ionizable starting gas such as argon at a pressure of 4 torrs.

10 The spectral power distribution for a bare high-pressure mercury lamp is shown in Fig. 2 for the green and yellow regions of the spectrum. The primary visible emissions for this lamp occur as a green line centered at about 546 nm and two yellow lines centered at 15 about 577 nm and 579 nm. A layer 30 comprising selected inorganic phosphor is carried on the inner surface of the protective outer envelope 14 of the lamp as shown in Fig. 1. A typical phosphor coating weight is two mg/cm². The phosphor is selected to be absorptive for radiations in 20 the wavelength range of from 565 nm to 595 nm and transmissive for visual radiations shorter than 565 nm and longer than 595 nm, and the phosphor also converts a substantial proportion of absorbed radiations in the wavelength range of from 565 nm to 595 nm to longer wavelength emissions in the range of from 596 nm to 640 nm. 25 In this manner, yellow emissions which are deleterious to a preferred color rendition can be converted to more desirable red emissions, in order to improve the perceived coloration of objects illuminated by the composite mercury lamp.

30 A variety of phosphors can be used for the foregoing application and in the following Table I are listed five known phosphors which will satisfy the foregoing requirements.

TABLE I

	Gadolinium antimonate:Eu	$(Gd_{0.9}Eu_{0.1}SbO_4)$
	Gadolinium tungstate:Eu	$(Gd_{1.98}Eu_{0.02}WO_6)$
	Strontium-sodium titanate:Eu	$(Sr_{1.6}Eu_{0.2}Na_{0.2}TiO_4)$
5	Calcium-yttrium silicate:Eu	$[Ca_2Y_{7.98}Eu_{0.02}(SiO_4)_6O_2]$
	Caldium-Lantuanum silicate:Eu	$[Ca_2La_{7.98}Eu_{0.02}(SiO_4)_6O_2]$

The foregoing phosphors are all responsive to certain wavelengths of ultraviolet excitation in order to produce the desired red radiations and they also all have an absorption band in the yellow region of the visible spectrum and convert absorbed yellow radiations into the specified desired reddish radiations. To illustrate, the spectral energy distribution for the emission of strontium-sodium titanate activated by trivalent europium is set forth in Fig. 3. In taking this curve, the phosphor was excited by 360-370 nm ultraviolet and the resolution of the apparatus was one nm. In Fig. 4 is shown the room temperature excitation spectrum of the so-called $^5D_0 \rightarrow ^7F_2$ transition red fluorescence of trivalent europium in the same strontium-sodium titanate matrix, wherein the resolution of the equipment used in taking the curve was four nanometers. Thus, by using this phosphor as a coating, yellow emissions from the mercury discharge are absorbed and converted to desired red emissions.

In Fig. 5 is plotted color-rendering index versus wavelength for each of three emissions combined to make white light. As is well established, the internationally accepted method for standardizing and measuring the color-rendering index of light sources is set forth in the publication of the International Commission on Illumination, identified as Publication C.I.E. No. 13 (E-1.3.2) 1965. The data reported in Fig. 5 are also disclosed in JOSA, Volume 61, No. 9 pp. 1155-1163 (September 1971), article referred to in the Background section. In explanation of Fig. 5, the curves were prepared by utilizing three optimum emissions, namely 450 nm, 540 nm and 610 nm.

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Two of these three optimum emissions were maintained at the foregoing specified wavelengths and the third was varied over a wide range to show the effect on color rendition, with the composite color of the resulting blend 5 of radiations always maintained the same to produce so-called "daylight" white light. By way of further explanation, in Fig. 5 the curve designated "B" for blue was obtained by maintaining the emissions of 540 nm and 610 nm constant, while varying the short wavelength "B" radiations from 410 nm to approximately 500 nm. The curve 10 designated "G" for green represents the results obtained by maintaining the 450 nm and 610 nm radiations constant while varying the middle or "G" radiations, and the curve "R" for red was similarly taken. As shown therein, by 15 minimizing the yellow-appearing radiations and maximizing the other desired radiations, including red radiations of the specified wavelengths, the color rendering capabilities of the light sources are enhanced. In addition, the so-called clarity of vision can be improved as well as the 20 apparent brightness of the objects illuminated.

In Figure 6 is shown the standard photopic curve as a dashed line and this curve is the standard which is used in rating the luminous efficacy for all light sources. This curve is also the standard used for all 25 other light-level measurements, such as are made using foot-candle meters. In accordance with this photopic curve, the eye is extremely sensitive to yellow-appearing radiations. In recent years, reports from scientists throughout the world have cast considerable doubt upon the 30 validity of the standard photopic curve and in Figure 6, superimposed over the dashed photopic curve, is a modified curve setting forth what is believed to be the true response of the human visual system to varying colored lights. As shown, there are three peaks occurring at 35 approximately 450 nm, 540 nm and 610 nm and two valleys occurring in the range of 490 nm (blue-green) and 575 nm (yellow). Thus, by converting the specified yellow-appearing radiations to more desirable red radiations, not

only can the color rendering properties of the light sources be improved, but if this modified curve ultimately proves to be valid, it will mean that the visual system actually responds to or "sees" the reddish radiations 5 better than the yellow radiations. It is pointed out, however, that as of this date, the photopic curve still remains the international standard for rating the luminous efficacy of light sources.

In Figure 7 is shown a fragmentary enlarged sectional view of an alternative embodiment for a portion of the protective outer envelope 14 of the lamp embodiment shown in Figure 1. Two separate layers of phosphor are carried on the interior surface of the envelope 14 with the phosphor layer 30 positioned next to the envelope 10 interior surface being any one or a mixture of the phosphors listed under the foregoing Table 1. The phosphor layer 32 which is carried over the layer 30 is positioned 15 to be nearer the discharge and the phosphor of the layer 32 is also used for purposes of color correcting the 20 discharge. An example of such phosphor is blue-emitting alkaline-earth metal halophosphate activated by divalent europium as disclosed more specifically in U.S. Patent No. 4,038,204 (Wachtel). In such an embodiment, the blue-emitting phosphor layer 32 which is nearest the discharge 25 intercepts a substantial proportion of the ultraviolet radiations and transmits the green and yellow visible radiations emitted by the high pressure mercury discharge. Thus yellow radiations are intercepted by the phosphor layer 30 for conversion to the desired red radiations.

Another alternative embodiment is shown in 30 Figure 8 wherein an additional layer 34 of extremely finely divided light-scattering material such as silica is deposited directly on the envelope interior surface, with the yellow to red converting phosphor layer 30 coated 35 thereover and the blue fluorescing layer 32 coated over the layer 30 to lie nearest the discharge. In such an embodiment, some of longer wavelength radiations, such as yellow radiations which pass the phosphor layer 30 without

being absorbed, are scattered back through the layer 30 so that the composite structure has enhanced absorption characteristics for the yellow radiations for conversion to the desired red radiations. As another possible embodiment, the phosphor layer 30 can have silica mixed therewith, or the phosphors of the layers 30 and 32 can themselves be mixed.

In Figure 9 is shown a high pressure sodium vapor lamp 36 which is modified in accordance with the present invention by carrying on the inner surface of the protective envelope 38 a layer 40 of any of the phosphors, or any mixture thereof, as listed under the foregoing Table 1. Other than the phosphor layer 40, the lamp is generally conventional and comprises an elongated arc tube 42 formed of alumina and having electrodes 44 operatively positioned therein proximate the ends thereof. Lead-in conductors 46, 48 connect to the lamp electrodes through the ends of the arc tube 42 and include a conventional supporting frame 50 which supports the arc tube in predetermined position within the lamp envelope. Connected to the frame 50 is a flexible lead-in conductor 52 to permit expansion and contraction of the arc tube 42 and conventional spring spacers 54 support the frame within the dome portion of the protective envelope 38. To complete the description, the outer envelope normally encloses a hard vacuum which is obtained through flashing getters from suitable getter holders 56. The lead-in conductors 46, 48 are sealed through a conventional stem-press arrangement 58 and connect to the conventional screw-type base 60. In accordance with the present invention, some of the yellow-appearing radiations emitted by the high-pressure sodium discharge are converted to red-appearing radiations by the phosphors as specified.

In Fig. 10 is shown a so-called mercury-metal halide HID lamp 62 modified in accordance with the present invention by incorporating as a layer 64 on the envelope inner surface any of the phosphors, or mixtures thereof, as specified under the foregoing Table 1. Apart from the

phosphor layer, the lamp construction is generally conventional and such a lamp is disclosed in U.S. Patent No. 3,798,487 (Zollweg et al). Briefly, the lamp comprises the arc tube 66 which encloses any of a variety of combinations of mercury plus metal halide 68, as is well-known in the art, operating electrodes 70, starting electrode 72, arc tube supporting frame 74, starting resistor 76, conare sealed to connect to the screw-type base 82. The functioning of the phosphor layer 64 is to convert yellow-appearing radiations to the specified red-appearing radiations, to enhance the perceived coloration of illuminated objects.

The present phosphor can also be incorporated as a part of a fluorescent lamp and such an embodiment is shown in Figure 11, wherein the lamp 82 has the yellow to red converting phosphor carried as a layer 84 proximate the lamp envelope 85 with a layer of conventional halophosphate phosphor 86 coated thereover. In the operation of the lamp, the low-pressure mercury discharge excites the halophosphate phosphor to emit visible radiations, which normally are relatively rich in yellow-appearing radiations. These yellow radiations then excite the phosphor layer 84 to emit the desired red-appearing radiations. The lamp 82 is otherwise conventional and comprises electrodes 88 operatively positioned proximate the ends of the elongated envelope, with conventional stem-press arrangements 90 through which lead-in conductors 92 are sealed to connect to conventional base caps 93 positioned at the ends of the envelope.

In any of the embodiments shown in Figures 9 and 10, the phosphors as listed under Table I can be used alone as a separate layer, or mixed with other color-correcting or converting phosphors, or coated as separate layers with or without additional light scattering materials such as shown in Fig. 8. In the case of a fluorescent lamp, it is normally desirable to use the yellow-to-red converting phosphors as a separate layer with the conventional phosphors, such as the well-known halophos-

phates, coated thereover. Even in this case, however, the yellow-to-red converting phosphor and the halophosphates may be mixed and coated as a single layer, if desired.

As another possible embodiment, in Figure 12 is shown a standard incandescent lamp which is modified to have coated over the envelope interior surface a layer of any of the phosphors which are specified under Table I. In addition, a layer of light scattering material may be carried next to the envelope and the phosphor coated thereover in order to enhance the absorption of yellow radiations from the emissions of the incandescent filament as normally operated. Other than the phosphor layer, the lamp 94 as shown in Figure 12 is generally conventional and comprises the protective outer envelope 96 having the phosphor 98 coated as a layer on the inner surface thereof, and an incandescent filament 100 supported in predetermined position within the outer envelope 96 by a pair of lead-in conductors 102 which connect through a conventional stem-press 104 to a conventional screw-type base 106.

In all of the foregoing embodiments, the light source comprises a source of radiations including a light-transmitting envelope with the light source as normally operated having appreciable emissions in the wavelength range of from 565 nm to 595 nm plus substantial emissions of visible radiations of wavelengths shorter than 565 nm and longer than 595 nm. These light sources are modified by incorporating therein a layer comprising inorganic phosphor which is carried on the light-transmitting envelope with the phosphor being absorptive for radiations in the wavelength range of from 565 nm to 595 nm and transmissive for visible radiations shorter than 565 nm and longer than 595 nm. The phosphor is operable to convert a substantial proportion of absorbed radiations in the wavelength range of from 565 nm to 595 nm to longer wavelength emissions in the range of from 596 nm to 640 nm in order to enhance the perceived coloration of objects which are illuminated by the light source.

The yellow-to-red converting phosphor preferably is in finely divided form, such as the two to ten micron-size range. The phosphor could be formed as a thin film, however, if desired. To protect the phosphor coating from abrasion, it is desirable to coat same on the inner surface of the protective envelopes. Since a yellow-to-red conversion is involved, however, the phosphor could also be coated or carried on the outer surface of the protective envelope.

What we claim is:

1. A light source for illuminating objects with an enhanced perceived coloration which comprises an electrically energizable source of radiations surrounded by a light-transmitting envelope, said source of radiations when energized having appreciable emissions in the wavelength range of from 565 nm to 595 nm plus substantial emissions of visible wavelengths shorter than 565 nm; characterized in that a layer comprising inorganic phosphor is carried on the inner surface of the light-transmitting envelope, said phosphor being absorptive for radiations in the wavelength range of from 565 nm to 595 nm and transmissive for visible radiations shorter than 565 nm and longer than 595 nm, and said phosphor layer being able to convert a substantial proportion of absorbed radiations in the range of from 565 nm to 595 nm to longer wavelength radiations in the wavelength range of from 596 nm to 640 nm.

2. A light source according to claim 1, characterized in that said light source is an incandescent lamp, and said source of radiations is an incandescent filament.

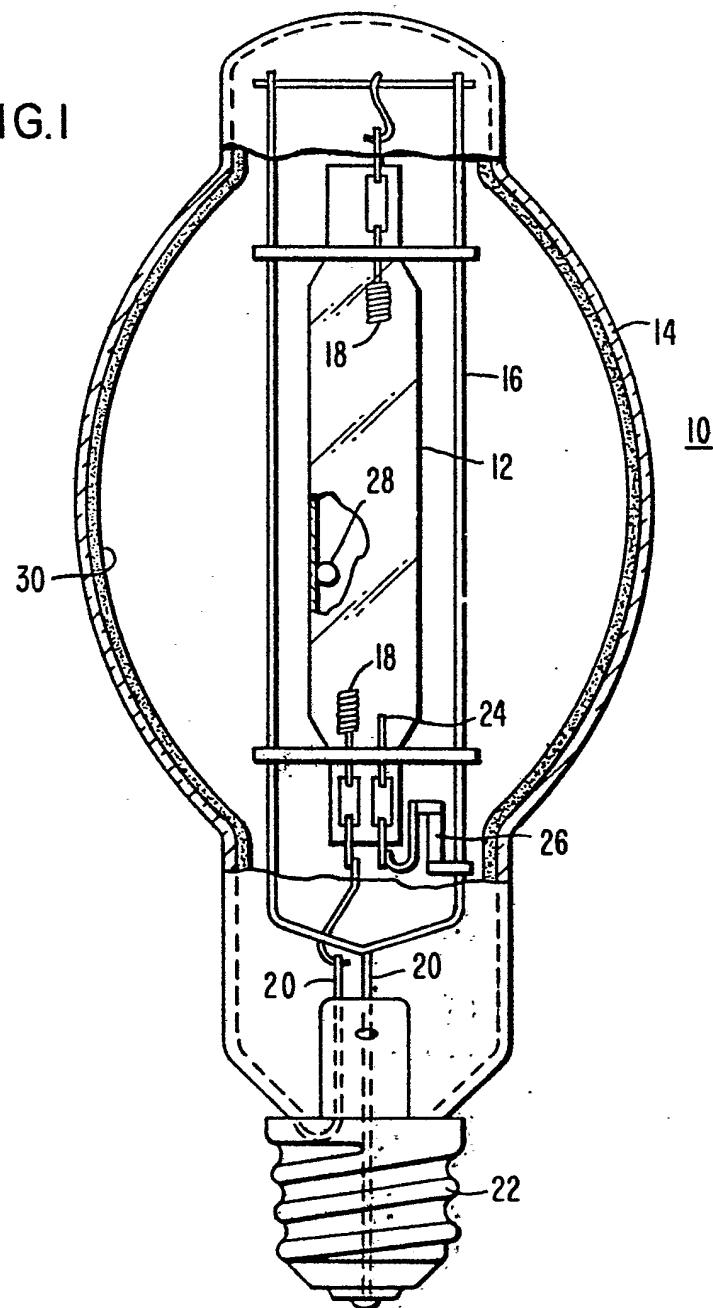
3. A light source according to claim 1, characterized in that the light source is a fluorescent lamp of the low pressure mercury discharge type, and the envelope has an elongated configuration with the phosphor being in finely divided form and carried on the inner surface thereof.

4. A light source according to claim 1, charac-

terized in that the light source is a high-intensity-discharge lamp which comprises a protective light-transmitting envelope, an arc tube supported within said protective envelope and adapted to have a discharge maintained therein to generate the energized source of radiations, and electrical lead-in means sealed through said protective envelope for connecting said arc tube to an energizing power source.

5 10 15 20 25 30 35 40 45 50 55 60 65 70 75 80 85 90 95 100 105 110 115 120 125 130 135 140 145 150 155 160 165 170 175 180 185 190 195 200 205 210 215 220 225 230 235 240 245 250 255 260 265 270 275 280 285 290 295 300 305 310 315 320 325 330 335 340 345 350 355 360 365 370 375 380 385 390 395 400 405 410 415 420 425 430 435 440 445 450 455 460 465 470 475 480 485 490 495 500 505 510 515 520 525 530 535 540 545 550 555 560 565 570 575 580 585 590 595 600 605 610 615 620 625 630 635 640 645 650 655 660 665 670 675 680 685 690 695 700 705 710 715 720 725 730 735 740 745 750 755 760 765 770 775 780 785 790 795 800 805 810 815 820 825 830 835 840 845 850 855 860 865 870 875 880 885 890 895 900 905 910 915 920 925 930 935 940 945 950 955 960 965 970 975 980 985 990 995 1000 1005 1010 1015 1020 1025 1030 1035 1040 1045 1050 1055 1060 1065 1070 1075 1080 1085 1090 1095 1100 1105 1110 1115 1120 1125 1130 1135 1140 1145 1150 1155 1160 1165 1170 1175 1180 1185 1190 1195 1200 1205 1210 1215 1220 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10185 10190 10195 10200 10205 10210 10215 10220 10225 10230 10235 10240 10245 10250 10255 10260 10265 10270 10275 10280 10285 10290 10295 10300 10305 10310 10315 10320 10325 10330 10335 10340 10345 10350 10355 10360 10365 10370 10375 10380 10385 10390 10395 10400 10405 10410 10415 10420 10425 10430 10435 10440 10445 10450 10455 10460 10465 10470 10475 10480 10485 10490 10495 10500 10505 10510 10515 10520 10525 10530 10535 10540 10545 10550 10555 10560 10565 10570 10575 10580 10585 10590 10595 10600 10605 10610 10615 10620 10625 10630 10635 10640 10645 10650 10655 10660 10665 10670 10675 10680 10685 10690 10695 10700 10705 10710 10715 10720 10725 10730 10735 10740 10745 10750 10755 10760 10765 10770 10775 10780 10785 10790 10795 10800 10805 10810 10815 10820 10825 10830 10835 10840 10845 10850 10855 10860 10865 10870 10875 10880 10885 10890 10895 10900 10905 10910 10915 10920 10925 10930 10935 10940 10945 10950 10955 10960 10965 10970 10975 10980 10985 10990 10995 11000 11005 11010 11015 11020 11025 11030 11035 11040 11045 11050 11055 11060 11065 11070 11075 11080 11085 11090 11095 11100 11105 11110 11115 11120 11125 11130 11135 11140 11145 11150 11155 11160 11165 11170 11175 1

FIG. I



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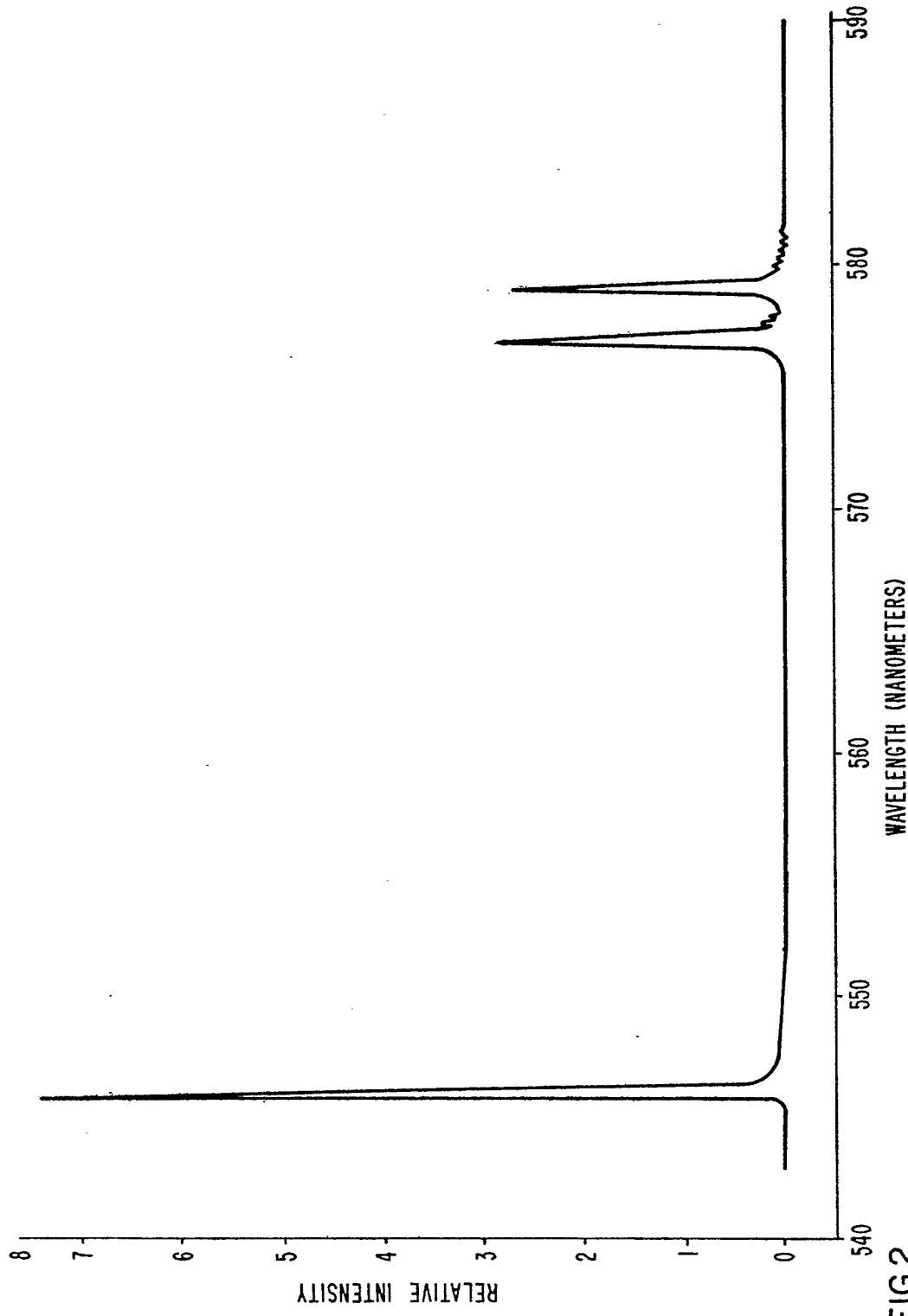


FIG.2

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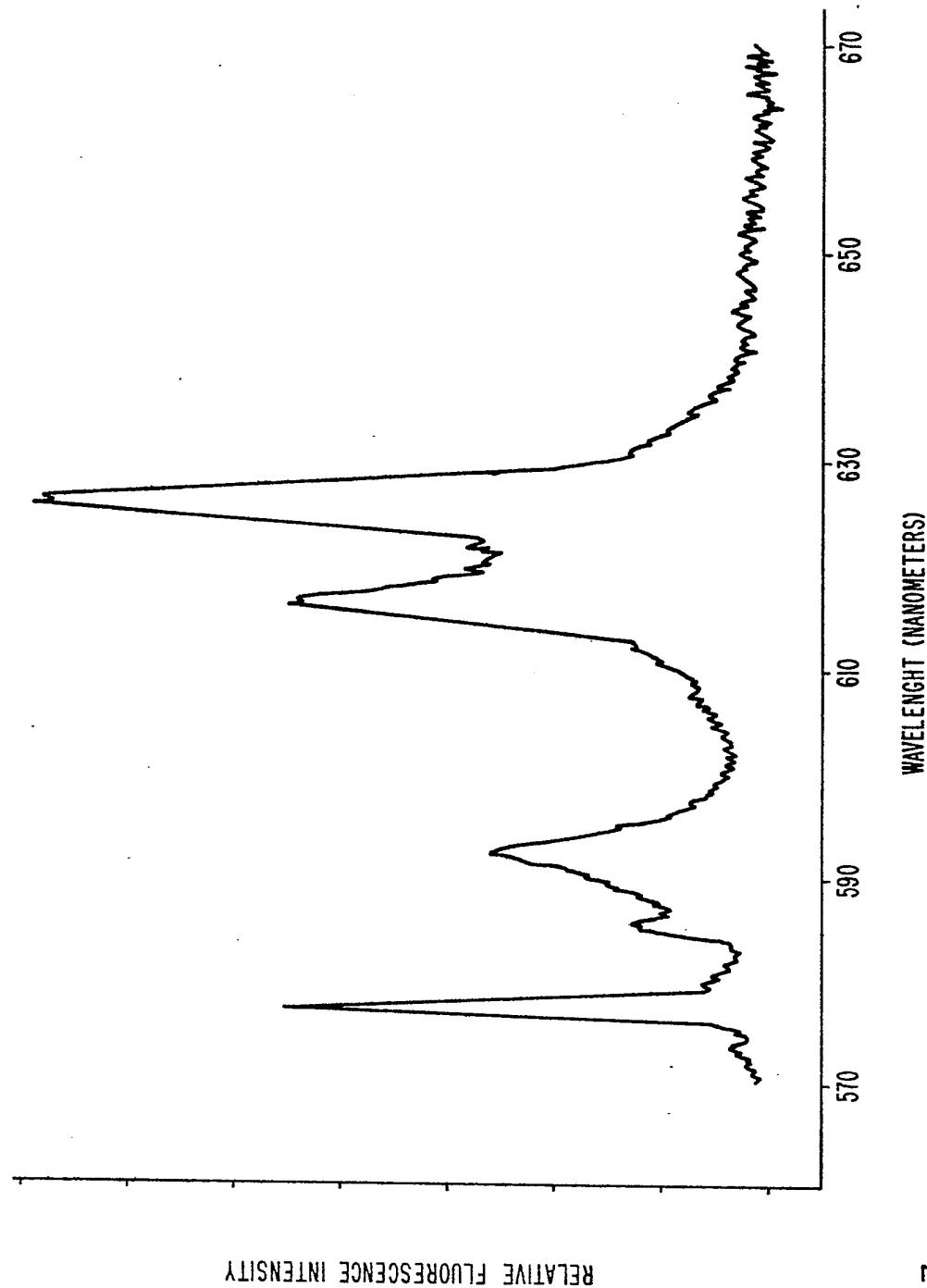


FIG.3

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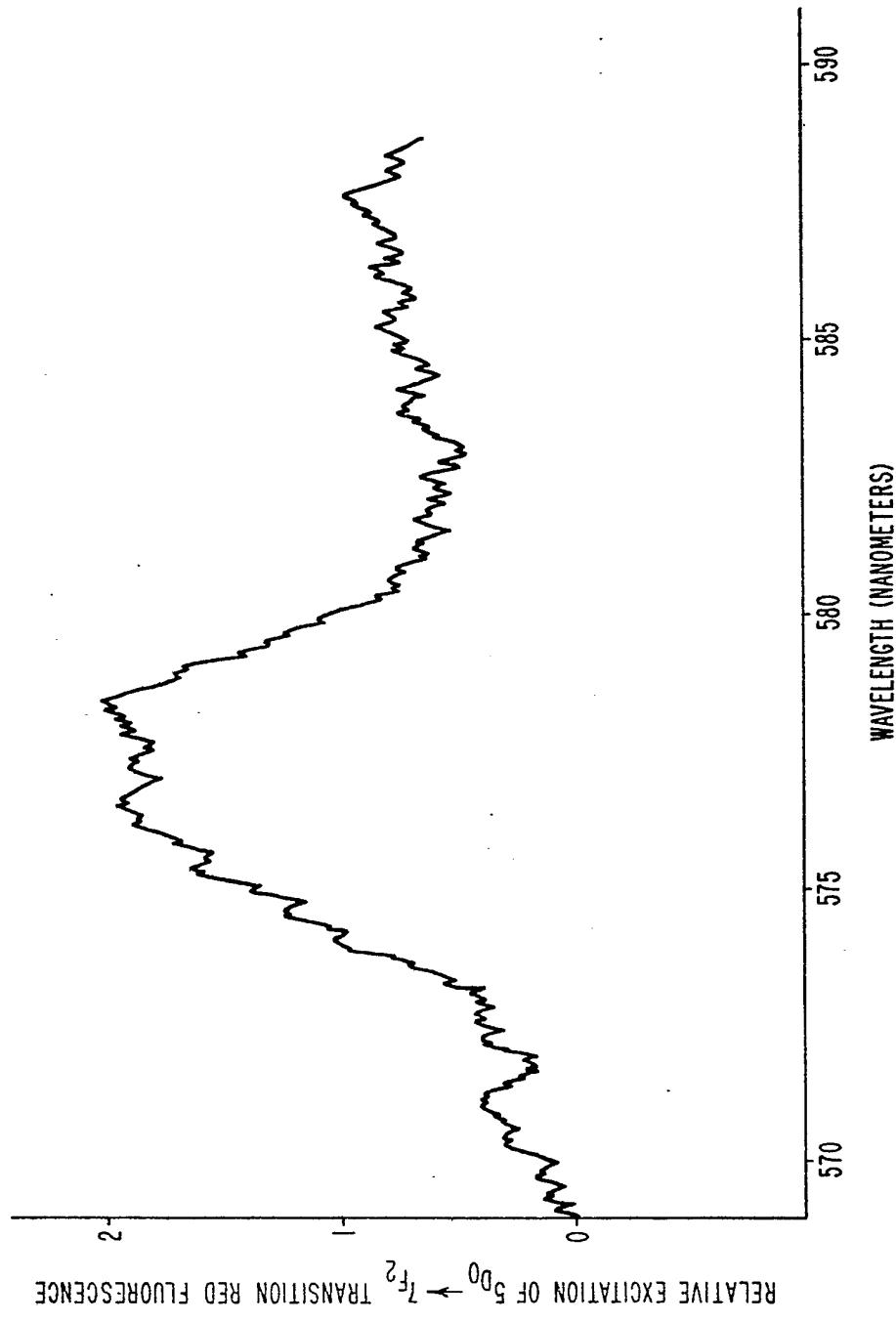


FIG.4

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FIG.5

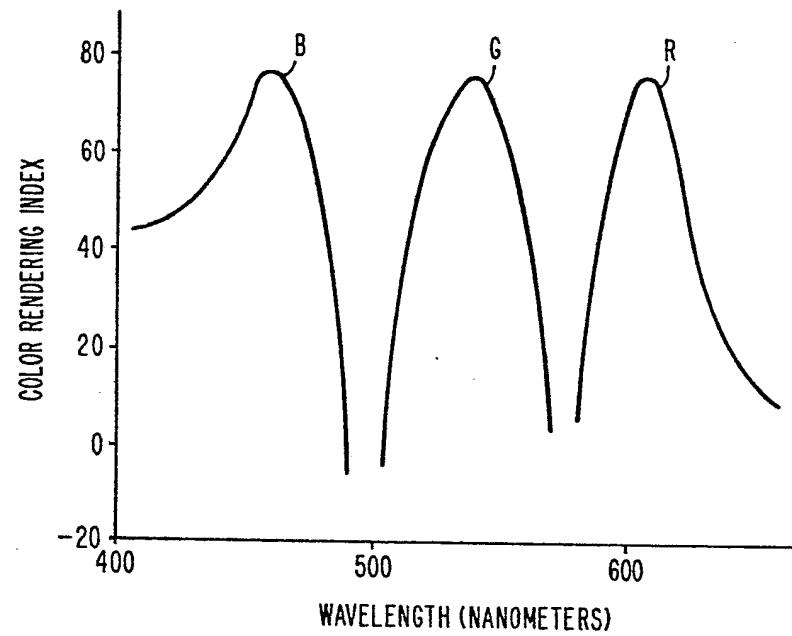
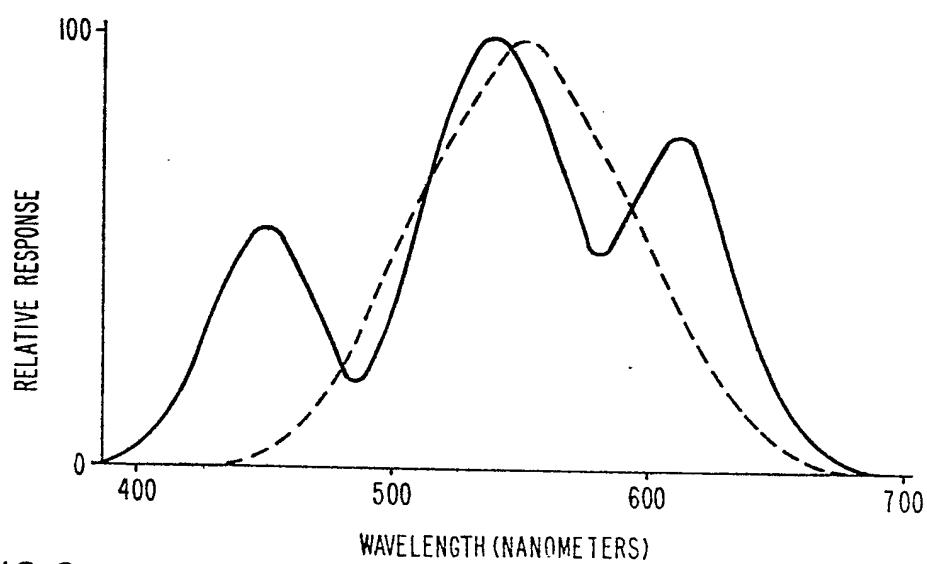


FIG.6



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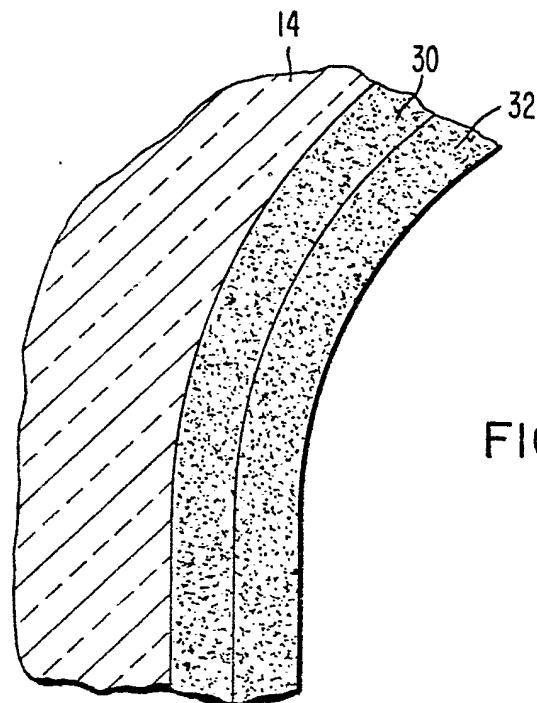


FIG.7

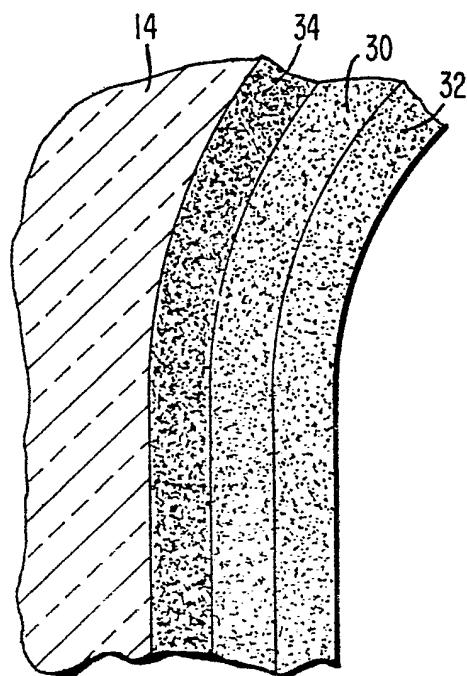
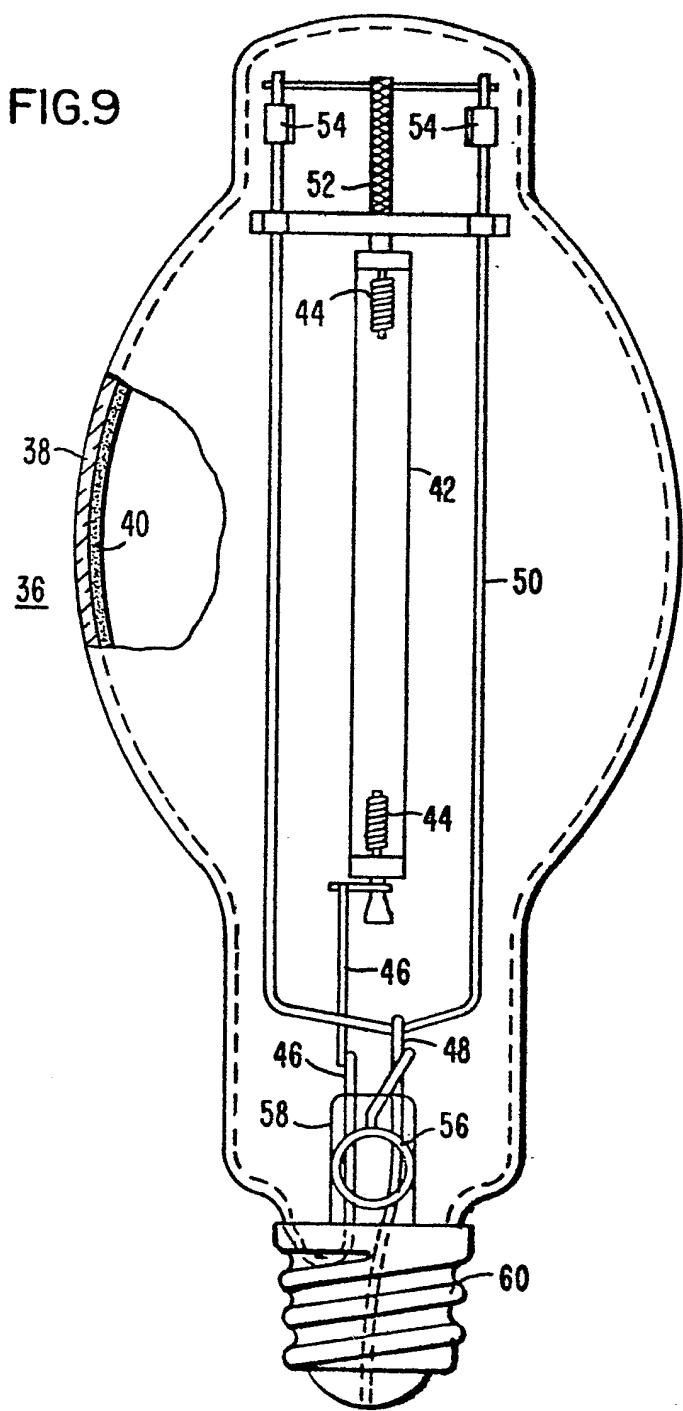


FIG.8

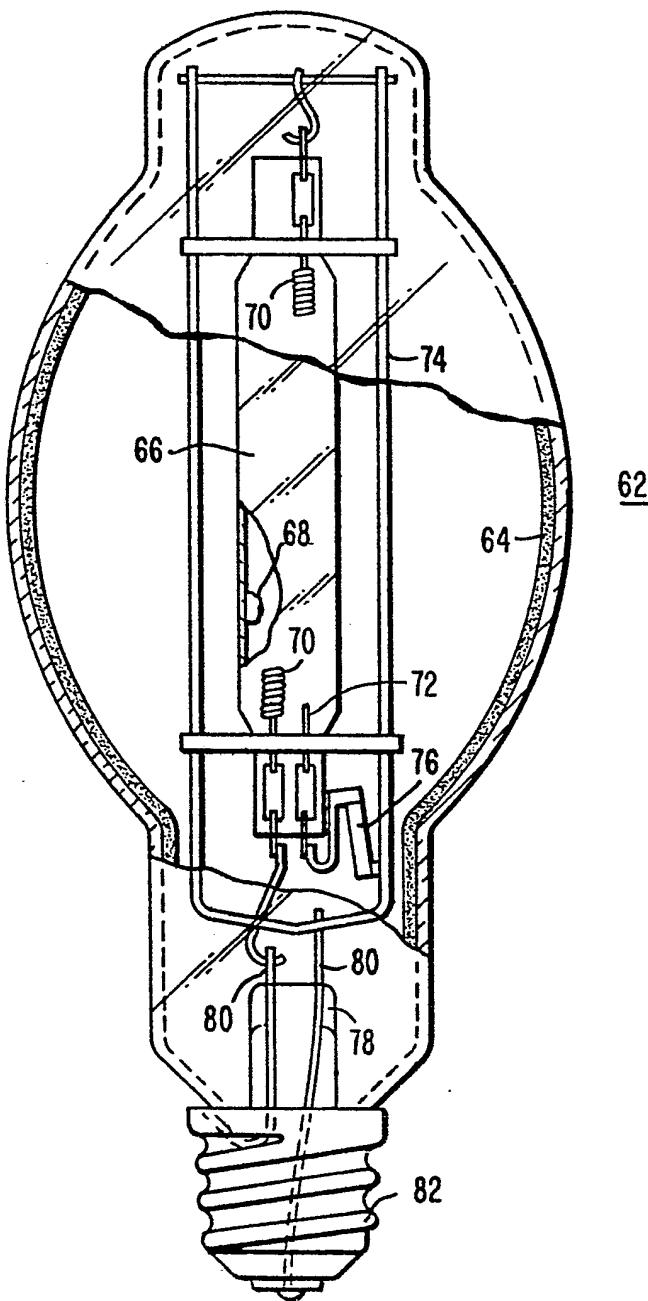
FIG.9



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FIG.10



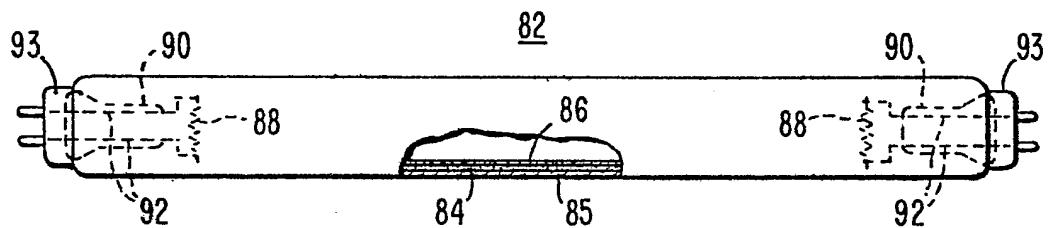


FIG.11

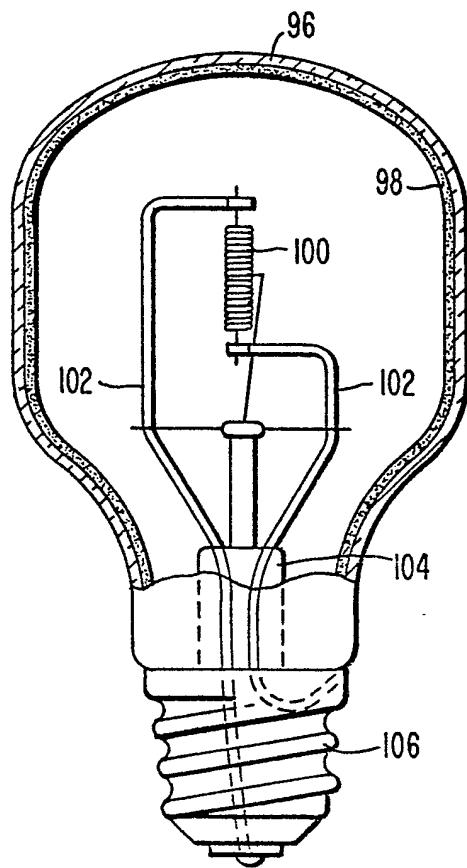


FIG.12